International Chemical Safety Cards

MERCURY

MERCURY Quicksilver Liquid silver Hydrargyrum Hg							
CAS # 7439-97-6 RTECS # OV455 ICSC # 0056 UN # 2809 EC # 080-001-00	0000	Ate	omic mass: 200.6				
TYPES OF HAZARD/ EXPOSURE	ACUTE HAZARDS/ SYMPTOMS		PREVENTION		FIRST AID/ FIRE FIGHTING		
FIRE			NO contact with flammable substances.		In case of fire in the surroundings: all extinguishing agents allowed.		
EXPLOSION	Risk of fire and explosion on contact with incompatible substances (see Chemical Dangers).				In case of fire: keep drums, etc., cool by spraying with water.		
EXPOSURE			STRICT HYGIENE! AVOID EXPOSURE OF (PREGNAN WOMEN! AVOID EXPOSUF ADOLESCENTS AND CHILDREN!		IN ALL CASES CONSULT A DOCTOR!		
• INHALATION			Local exhaust or breathing protection.		Fresh air, rest. Artificial respiration if indicated. Refer for medical attention.		
• SKIN	MAY BE ABSORBED!		Protective gloves. Protective clothing.		Remove contaminated clothes. Rinse and then wash skin with water and soap. Refer for medical attention.		
• EYES			Face shield, or eye protection in combination with breathing protection.		First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then take to a doctor.		
• INGESTION		Do not eat, drink, or smoke during work. Wash hands before eating.		Refer for medical attention.			
SPILLAGI	E DISPOSAL		STORAGE	PA	CKAGING & LABELLING		
Ventilation. Collect leaking and spilled extinguishing liquid in sealable non-metallic containers as		. Separated from azides, monia, food and feedstuffs. Ventilation along the floor. U		pecial material. Do not transport with food nd feedstuffs. 'symbol :: 23-33 : (1/2-)7-45 JN Hazard Class: 8 JN Packing Group: III			
SEE IMPORTANT INFORMATION ON BACK							
ICSC: 0056 Prepared in the context of cooperation between the International Programme on Chemical Safety & the Commission of the European Communities © IPCS CEC 1993							

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I	PHYSICAL STATE; APPEARANCE: ODOURLESS, HEAVY AND MOBILE SILVERY	ROUTES OF EXPOSURE: The substance can be absorbed into the body by				
М	LIQUID METAL.	inhalation and through the skin, also as a vapour!				
P		INHALATION RISK: A harmful contamination of the air can be reached				
0	CHEMICAL DANGERS:	very quickly on evaporation of this substance at 20°C.				
R	Upon heating, toxic fumes are formed. Reacts violently with alkali metals, acetylene, azides,	EFFECTS OF SHORT-TERM EXPOSURE: Inhalation of the vapours may cause pneumonitis. The				
Т	ammonia gas, chlorine, chlorine dioxide, sodium carbide and ethylene oxide. Attacks copper and many other metals forming amalgams.	substance may cause effects on the kidneys and the central nervous system. The effects may be delayed. Medical observation is indicated.				
Α	OCCUPATIONAL EXPOSURE LIMITS (OELs):	EFFECTS OF LONG-TERM OR REPEATED				
Ν	TLV: ppm; 0.025 mg/m ³ (as TWA) (skin) (ACGIH	EXPOSURE:				
Т	1994-1995). MAK: 0.01 ppm; 0.1 mg/m ³ ; (1992).	The substance may have effects on the central nervous system and kidneys, resulting in emotional and psychic instability, tremor mercurialis, cognitive disturbances, speech disorders. Danger of cumulative				
D		effects. Animal tests show that this substance possibly causes toxic effects upon human reproduction.				
Α						
Т						
Α						
PHYSICAL PROPERTIES	Boiling point: 357°C Melting point: -39°C Relative density (water = 1): 13.5 Solubility in water: none	Vapour pressure, Pa at 20°C: 0.26 Relative vapour density (air = 1): 6.93 Relative density of the vapour/air-mixture at 20°C (air = 1): 1.009				
ENVIRONMENTAL DATA The substance is very toxic to aquatic organisms. In the food chain important to humans, bioaccumulation takes place, specifically in fish.						
	N O T E S					
Depending on the degree of exposure, periodic medical examination is indicated. No odour warning if toxic concentrations are present. Do NOT take working clothes home.						
ADDITIONAL INFORMATION						
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